THOMAS ADEWUMI UNIVERSITY					
COURSE OUTLINE					
Faculty	Computing and Applied Sciences				
Department	Biological Sciences				
Course Title	GENERAL BIOCHEMISTRY PRACTICAL I				
Year of Study	2				
Course Code	BCH 205				
Credit Hours	1				
Contact Hours	15				
Mode of Delivery	Classroom Lectures				
	Laboratory Practical Sessions				
Mode of Assessment		Weight %			
Continuous Assessment/Practical sessions		50%			
Final Examination		50%			
Total		100 %			
Course Lecturers	FAROHUNBI S.T.				
Lab Instructor	Miss Onukogu Stella				
Course Description	The course is expected to expose the students to Laboratory experiments to reflect the topics covered in BCH 201. Isolation techniques: Homogenization and centrifugation. Measurement of pH, buffers; acidity and alkalinity; pH and pKa values.				
Course objective	This course would enable the understanding practiful following:	ical aspects of the			
	 Laboratory ethics and practices in Biochemistry and biosciences Homogenization and centrifugation 				

	2 pH and Puffors				
	3. pH and Buffers4. acidity and alkalinity				
	5. homogenization and centrifugation				
	6. extraction and isolation of proteins from plant/animal sources				
Learning Outcomes	By the end of the course, student will be able to carry out 1. Homogenization and centrifugation				
	2. pH and Buffers				
	3. acidity and alkalinity				
	4. homogenization and centrifugation5. extraction and isolation of proteins from plant/animal source				
Teaching and	The class will meet for two hours each week. Class time will be used				
Learning	for a combination of lectures and Tutorial sessions				
Dear ming	Tot a combination of fectures and 1 atorial sessions				
Detailed Course	Intracellular localization of enzymes. Properties of enzymes. Enzyme				
Content	kinetic and inhibition; co-enzymes and cofactors. Glycolysis,				
	Tricarboxylic acid cycle, Oxidative Phosphorylation and Hexose				
	monophosphate shunt. Membranes and transport. Glycogen synthesis				
	and breakdown. Oxidative deamination, transamination, and urea				
	cycle. Degradation of amino acid. Synthesis of fatty acids, oxidation				
	of fatty acids. DNA replication and transcription: protein biosynthesis				
	and regulation. Cholesterol: chemistry, synthesis, and breakdown.				
	Biochemical basis of hormone action. Drug metabolism. Mineral metabolism and role of calcium in bone formation. Introduction to				
	Nutritional Biochemistry.				
	Course content sequencing				
Weeks	Detailed Course Outline	Allocated			
		Time			
Week 1-2	Intracellular localization of enzymes	4 hours			
	Properties of enzymes				
	Enzymes kinetic and inhibition				
Week 3-4	Coenzymes and Cofactors	4 hours			
	Glycolysis, Tricarboxylic acid cycle				
	Oxidative phosphorylation and Hexose				
	monophosphate shunt				
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Week 5-7	Membranes and Transport	6 hours
	Glycogen synthesis and breakdown Oxidative deamination, transamination and urea	
	cycle Degradation of amino acid, Synthesis of fatty acids, oxidation of fatty acids	
Week 8-9	DNA replication and transcription; protein biosynthesis and regulation Cholesterol: chemistry, synthesis and breakdown	4 hours
Week 10	Biochemical basis of hormone action Drug metabolism. Mineral metabolism and role of calcium in bone formation. Introduction to Nutritional Biochemistry	
After Week 12	Examinations	

Recommended Reading Material

- 1. David, L., Nelson, D.L., Cox, M.M., Stiedemann, L., McGlynn Jr, M.E. and Fay, M.R., 2000. Lehninger principles of biochemistry.
- 2. Lieberman, M. and Marks, A.D., 2009. *Marks' basic medical biochemistry: a clinical approach*. Lippincott Williams & Wilkins.
- 3. Rodwell, V.W., 2015. Harper's illustrated biochemistry. McGraw-Hill Education.
- 4. Vasudevan, D.M., Sreekumari, S. and Vaidyanathan, K., 2019. *Textbook of biochemistry for medical students*. Jaypee Brothers Medical publishers.
- 5. Chatterjea, M.N. and Shinde, R., 2011. *Textbook of medical biochemistry*. Life Goes On.